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A New Tactic for Diastereo- and Enantiocontrolled Synthesis of (-)-Malyngolide via Catalytic Meso-Asymmetrization

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Abstract: A meso-1,4-enediol bis-silyl ether 2 having bicyclo[2.2.1]heptene background has been transformed diastereo- and enantioselectively into (-)-malyngolide 1, an antibiotic isolated from the blue-green marine algae, Lyngbya majuscula, via Rh(I)-(R)BINAP-catalyzed asymmetrization and diastereoselective modification of the optically active product thus obtained. © 1997 Elsevier Science Ltd.

Despite its simple structure having one quaternary and one tertiary stereogenic center on the δ -lactone ring, diastereocontrolled construction of the two centers of (-)-malyngolide¹ 1, an antibiotic isolated from the blue-green marine algae, Lyngbya majuscula, is not an easy task. Although nearly two dozen methods for its chiral synthesis have so far been reported,² only a single procedure originally developed by Matsuo and Tanaka³ in their racemic synthesis has succeeded in installing the two stereogenic centers with satisfactory diastereoselectivity.⁴ Thus, two stereogenetic centers were first installed enantio- and diastereoselectively on the cyclopentanone framework and the chiral 2,2,3-trisubstituted cyclopentanone generated was next transformed into (-)-malyngolide 1 by regioselective Baeyer-Villiger oxidation.^{2j,20,2p,21} We report herein a new tactic leading to high enantio- and diastereocontrolled construction (>96% ee:>99% de) of (-)-malyngolide 1 starting from the *meso-*1,4-enediol *bis*-silyl ether⁵ 2 having bicyclo[2,2,1]heptene background by use of a [Rh{(R)-BINAP}(COD)]ClO₄ [Rh(I)-(R)BINAP]⁶-catalyzed *meso*-asymmetrization reaction⁵ followed by diastereoselective modification of the optically active product based on its biased stereochemical background (Scheme 1).



Scheme 1

According to the established procedure,⁵ the tricyclic *meso-bis*-silyl ether 2 was refluxed in 1,2dichloroethane in the presence of a catalytic amount of Rh(I)-(R)BINAP (2 mol%) for 16 h to give the enol ether 3 which, immediately, was treated with tetrabutylammonium fluoride (TBAF) in THF (0 °C, 10 min) to afford the optically active keto ether 4, $[\alpha]_D^{30} + 26.6$ (c 1.5, CHCl₃), in nearly quantitative yield. Desilylation of 4 with TBAF (THF, rt, 20 h) or with a hydrogen fluoride-acetonitrile complex⁷ (-30 °C, 24 h) gave the hydroxy ketone 5, $\left[\alpha\right]_{0}^{29}$ +136.2 (c 1.9, CHCl₃), excellently, whose benzoate was proved to be 98% ee by hplc analysis using chiral column (CHIRALCEL OD, i-PrOH-hexane, 1:99 v/v). In order to construct the tertiary methyl group, 5 was treated with dibromomethane and zinc in the presence of titanium tetrachloride⁸ to give the diene 6, $[\alpha]_{D}^{25}$ +90.6 (c 1.3, CHCl₃), in 81% yield. When the methylenation was carried out in prior to the desilylation, the exo-methylene product was obtained in a much poorer yield. Exposure of the hydroxy olefin 6 to N-bromosuccinimide (NBS)⁹ in dichloromethane allowed chemoselective discrimination of two double bonds to give the single bromo-ether 7, $[\alpha]_{D}^{27}$ +176.4 (c 1.3, CHCl₃), in 88% yield leaving the exomethylene olefin. Hydrogenation of the exo-methylene 7 on Adams catalyst in methanol did not exhibit high convex-face diastereoselectivity which gave a mixture of endo- and exo-methyl isomers in 8:1 ratio. However, we were able to improve this less than satisfactory result by a fortuitous finding. Namely, when 7 was subjected to hydrogenation on 10% palladized carbon in dichloromethane, double bond migration instead of hydrogenation took place to generate the isomeric endo-olefin 8. On hydrogenation in methanol in the presence of Adams catalyst, the *endo*-olefin 8 afforded diastereoselectively the *endo*-methyl product 9, $[\alpha]_{D}^{31}$ +107.7 (c 0.9, CHCl₃), in an excellent yield (Scheme 2).



Having introduced the requisite tertiary methyl group, the bromo-ether 9 was next treated with zinc in boiling ethanol containing acetic acid⁹ so as to construct the quaternary center. The reaction proceeded readily as expected to give the secondary alcohol 10, mp 87-88 °C, $[\alpha]_D^{28} -2.7$ (*c* 1.3, CHCl₃), in 76% yield, which was oxidized with pyridinium dichromate (PDC) in dichloromethane to afford the ketone 11, $[\alpha]_D^{30} -36.5$ (*c* 0.5, CHCl₃). Grignard reaction of 11, though cerium trichloride⁹ was essential, proceeded diastereoselectively from the convex face to give the product 12, $[\alpha]_D^{28} -3.1$ (*c* 1.0, CHCl₃), having the requisite quaternary center in 77% yield. Thermolysis of the tertiary alcohol 12, in boiling diphenyl ether (~260 °C) in the presence of

sodium hydrogen carbonate,¹¹ initiated a retro-Diels-Alder reaction with some decomposition to afford the cyclohexenol 13, $[\alpha]_D^{28} + 27.7$ (*c* 0.3, CHCl₃), in 46% yield as a single isomer. On sequential ozonolysis and sodium borohydride reduction in the same flask, 13 furnished the triol 14, by cleavage of the olefin bond, which was immediately transformed to the acetonide 15, $[\alpha]_D^{28} - 5.1$ (*c* 0.7, CHCl₃), in 87% yield on reaction with 2,2-dimethoxypropane in the presence of pyridinium *p*-toluenesulfonate (PPTS).¹² Oxidation of the primary hydroxyl group of 15 with PDC in DMF gave the carboxylic acid 16 which, on sequential acid treatments, furnished the target molecule (-)-malyngolide 1, $[\alpha]_D^{28} - 13.4$ (*c* 0.3, CHCl₃) [natural¹: $[\alpha]_D - 13$ (*c* 2, CHCl₃)], in 53% overall yield *via* the glycol 17 by deacetonization and lactonization¹³ (Scheme 3).



In summary, by exploiting the Rh(I)-(R)BINAP-catalyzed *meso*-asymmetrization and diastereoselective modification based on the stereochemical background of the chiral product obtained, a new tactic for diastereoand enantiocontrolled construction of (–)-malyngolide 1 has been devised.

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REFERENCES AND NOTES

- 1. Cardillina, II, J. H.; Moore, R. E.; Arnold, E. V.; Clardy, J. J. Org. Chem. 1979, 44, 4039.
- (a) Sakito, Y.; Tanaka, S.; Asami, M.; Mukaiyama, T. Chem. Lett. 1980, 10, 1223. (b) Mukaiyama, T. Tetrahedron 1981, 37, 4111. (c) Pougny, J. -R.; Rollin, P.; Sinay, P. Tetrahedron Lett. 1982, 23, 4929. (d) Kogure, T.; Eliel, E. L. J. Org. Chem. 1984, 49, 576. (e) Noda, Y.; Kikuchi, M. Synth. Commun. 1985, 15, 1245. (f) Ho, P. T.; Wong, S. Can. J. Chem. 1985, 63, 2221. (g) Tokunaga, Y.; Nagano, H.; Shiota, M. J. Chem. Soc., Perkin Trans. 1 1986, 581. (h) Giese, B.; Rupaner, R. Liebigs Ann. Chem. 1987, 231. (i) Suemune, H.; Harabe, T.; Xie, Z. -F.; Sakai, K. Chem. Pharm. Bull. 1988, 36, 4337. (j) Sato, T.; Maeno, H.; Noro, T.; Fujisawa, T. Chem. Lett. 1988, 1739. (k) Trinh, M. -C.; Florent, J. -C.; Monneret, C. Tetrahedron 1988, 44, 6633. (l) Matsuo, K.; Hasuike, Y.;

Kado, H. Chem. Pharm. Bull. 1990, 38, 2847. (m) Ichimoto, I.; Machiya, K.; Kirihara, M.; Ueda, H. Agric. Biol. Chem. 1990, 54, 657. (n) Honda, T.; Imai, M.; Keino, K.; Tsubuki, M. J. Chem. Soc., Perkin Trans. 1 1990, 2677. (o) Guingant, A. Tetrahedron: Asymmetry 1991, 2, 415. (p) Asaoka, M.; Hayashibe, S.; Sonoda, S.; Takei, H. Tetrahedron 1991, 47, 6967. (q) Nagano, H.; Ohno, M.; Miyamae, Y.; Kuno, Y. Bull. Chem. Soc. Jpn. 1992, 65, 2814. (r) Ohira, S.; Moritani, M.; Ida, T.; Yamato, M. J. Chem. Soc., Chem. Commun. 1993, 1299. (s) Flörke, H.; Schaumann, E. Liebigs Ann. Chem. 1996, 147. (t) Enders, D.; Knopp, M. Tetrahedron 1996, 52, 5805.

- 3. Matsuo, K.; Tanaka, K. Chem. Pharm. Bull. 1981, 29, 3070.
- 4. There is another diastereo- and enantioselective synthesis which, however, was accomplished by combination of two chiral components, see: Ref. 2d.
- 5. Hiroya, K.; Kurihara, Y.; Ogasawara, K. Angew. Chem. Int. Ed. Engl. 1995, 34, 2287.
- A pertinent reviews, see: Otsuka, S.; Tani, K. In Asymmetric Synthesis Morrison, J. D. Ed., Academic Press, New York, 1985, Vol. 5, 171. Noyori, R.; Kitamura, M. Modern Synthetic Methods Schefford, E. Ed., Springer, Heidelberg, 1989, Vol. 5, 115.
- Newton, R. F.; Reynolds, D. P.; Finch, M. A. W.; Kelly, D. R.; Roberts, S. M. Tetrahedron Lett. 1979, 3981.
- 8. Takai, K.; Hotta, Y.; Oshima, K.; Nozaki, H. Bull. Chem. Soc. Jpn. 1980, 53, 1698.
- 9. Kamikubo, T.; Hiroya, K.; Ogasawara, K. Tetrahedron Lett., 1996, 37, 499.
- Imamoto, T.; Takiyama, N.; Nakamura, K.; Hatajima, T.; Kamiya, Y. J. Am. Chem. Soc. 1989, 111, 4392.
- 11. Kamikubo, T.; Ogasawara, K. J. Chem. Soc., Chem. Commun. 1995, 1951.
- 12. Miyashita, M.; Yoshikoshi, A.; Grieco, P. J. Org. Chem. 1977, 42, 3772.
- 13. Spectroscopic data of the representative intermediates: 8: ¹H NMR (CDCl₃, 300 MHz) δ=1.71 (3H, s), 2.14 (2H, m), 2.31 (3H, m), 2.55 (1H, m), 2.89 (1H, m), 3.65 (1H, d, J=2.2 Hz), 4.17 (1H, m), 4.47 (1H, d, J=5.2 Hz), 5.42 (1H, m). 9: ¹H NMR (CDCl₃, 300 MHz) δ=1.06 (3H, d, J=7.1 Hz), 1.38 (5H, m), 1.77 (1H, m), 1.93 (1H, m), 2.06 (1H, m), 2.12 (1H, m), 2.44 (1H, m), 2.87 (1H, m), 4.11 (1H, d, J=2.5 Hz), 4.19 (1H, m), 4.53 (1H, d, J=5.2 Hz), ¹³C NMR (CDCl₃, 75 MHz) δ=20.4, 25.0, 28.0, 31.0, 35.2, 38.4, 42.2, 45.2, 48.9, 57.0, 76.9, 88.8. 10: IR (film) $v=3266 \text{ cm}^{-1}$. ¹H NMR (CDCl₄, 300 MHz) δ =0.89 (3H, d, J=6.9 Hz), 0.97 (1H, ddd, J=3.6, 12.1, 12.1 Hz), 1.15 (1H, ddd, J=3.6, 12.1, 12.1 Hz), 1.27 (1H, d, J=8.0 Hz), 1.38 (1H, br s, exchangeable with D₂O), 1.45 (1H, dt, J=8.0, 1.7 Hz), 1.53 (1H, m), 1.75 (2H, m), 2.41 (1H, ddd, J=4.4, 5.2, 10.2 Hz), 2.60 (1H, ddd, J=3.6, 6.6, 10.2 Hz), 2.87 (1H, m), 2.97 (1H, m), 4.11 (1H, dt, J=12.1, 6.6 Hz), 6.09 (1H, dd, J=2.7, 5.8 Hz), 6.12 (1H, dd, J=2.7, 5.8 Hz). ¹³C NMR (CDCl₄, 75 MHz) δ=20.3, 26.6, 26.9, 27.1, 44.0, 44.5, 45.1, 45.3, 52.6, 68.4, 133.4, 133.6. 11: IR (film) v=1698 cm⁻¹. ¹H NMR (CDCl₃, 300 MHz) δ=1.11 (3H, d, J=6.9 Hz), 1.29 (1H, d, J=8.2 Hz), 1.39 (1H, dt, J=8.2, 2.2 Hz), 1.45 (2H, m), 2.10 (2H, m), 2.22 (1H, dt, J=17.3, 3.3 Hz), 2.65 (1H, ddd, J=3.0, 7.1, 9.3 Hz), 2.83 (1H, dd, J=4.4, 9.3 Hz), 3.03 (1H, m), 3.32 (1H, m), 6.06 (1H, dd, J=2.8, 5.5 Hz), 6.13 (1H, dd, J=2.8, 5.5 Hz). ¹³C NMR (CDCl₁, 75 MHz) δ=19.4, 26.7, 31.2, 40.4, 43.5, 45.2, 47.0, 49.8, 51.0, 135.5, 136.6, 214.1. 13: IR (film) $v=3372 \text{ cm}^{-1}$. ¹H NMR (CDCl₃, 300 MHz) $\delta=0.88$ (3H, t, J=6.6 Hz), 1.01 (3H, d, J=7.1 Hz), 1.10-1.60 (18H, m), 1.69 (2H, m), 1.96 (1H, br s, exchangeable with D₂O), 2.07 (1H, m), 5.57 (1H, dd, J=1.4, 9.9 Hz), 5.64 (1H, dd, J=6.3, 9.9 Hz). ¹³C NMR (CDCl₃, 75 MHz) δ =14.1, 21.2, 22.6, 23.6, 27.8, 29.3, 29.6, 30.2, 31.0, 31.9, 34.8, 42.5, 69.4, 132.0, 136.6.

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